## Catalytic Hydroboration

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## Copper(I)-Catalyzed Regioselective Monoborylation of 1,3-Enynes with an Internal Triple Bond: Selective Synthesis of 1,3-Dienylboronates and 3-Alkynylboronates\*\*

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Organoboron compounds are useful reagents and the hydroboration of simple alkenes or alkynes is one of the most efficient and straightforward methods to access a variety of organoboron compounds.<sup>[1,2]</sup> However, for the preparation of polyconjugated hydrocarbon compounds, there are limited types of regio- and stereoselective hydroboration reactions.<sup>[3-10]</sup> This type of transformation is still challenging in both transition-metal-catalyzed and noncatalyzed hydroboration

Hydroboration of 1,3-enyne compounds, for example, gives limited types of the organoboron products.<sup>[4-9]</sup> This transformation can theoretically produce six possible product isomers (Scheme 1, types **I–VI**). However, there has been no clear-cut report on the selective 1,2-hydroboration of 1,3-enynes (types **I** and **II**).<sup>[4,5]</sup> Allenylboron compounds can be obtained through palladium-catalyzed 1,4-hydroboration of

**Scheme 1.** Possible product isomers in the hydroboration of 1,3-enyne compounds.

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1,3-enynes (type III), [6] whereas the type IV product has not been reported. The 3,4-hydroboration of 1,3-enynes is the most common reaction pattern; the type V product, which is the 1,3-dienylboron compound, is a useful synthetic precursor. However, in this reaction the type VI product is detected as a minor product. Currently, the successful reaction patterns are limited to the production of type III and V products. In addition, most examples for type III and V products require the substrate structure to have a terminal alkyne moiety. [7-9] Although hydroboration is a general and widely used synthetic procedure, the application of hydroboration to 1,3-enynes, especially those with an internal alkyne moiety, remains undeveloped.

Very recently, our research group reported the copper(I)-catalyzed, regio- and enantioselective monoborylation of 1,3-diene compounds. This process can be extended for the development of novel regioselective borylation reactions of other conjugated systems; namely, where conventional hydroboration can not be used effectively. Herein, we report a copper(I)-catalyzed, highly regioselective monoborylation of 1,3-enyne compounds. In this catalysis reaction, either 3-alkynylboronates or 1,3-dienylboronates were obtained with high regioselectivity (Scheme 2). Substrates with a terminal double bond exclusively afforded unprecedented type I products (Scheme 2a), whereas highly substituted substrates gave type V product with high regioselectivity—even when the substrates have an internal alkyne moiety (Scheme 2c). Interestingly, in the reaction of 1,3-enynes that have moderate

a) Small steric hindrance around C=C

b) Moderate steric hindrance around C=C

c) Large steric hindrance around C=C

$$\begin{array}{c} R^2 \\ R^1 \\ \hline \\ R^3 \end{array} \\ \begin{array}{c} R^4 \\ \hline \\ B-B \end{array} \\ \begin{array}{c} \text{Cat. Cul/xantphos or PPh}_3 \\ \hline \\ B-B \end{array} \\ \begin{array}{c} R^2 \\ \hline \\ R^3 \\ \hline \\ R^4 \\ \hline \\ \text{type V} \end{array} \\ \end{array}$$

**Scheme 2.** Regioselective copper(I)-catalyzed monoborylation of 1,3-enyne compounds. THF = tetrahydrofuran, xantphos = 4,5-bis (diphenylphosphanyl)-9,9-dimethylxanthene.

steric demand around the double bond (Scheme 2b), ligandcontrolled regioselective borylation was observed. The synthetic utility of the reaction products was further demonstrated through the Suzuki-Miyaura cross-coupling and the Diels-Alder reaction. In addition, a preliminary result for the asymmetric 1,2-monoborylation of 1,3-enyne (84 % ee) is also reported.

The regioselectivity in the copper(I)-catalyzed monoborylation of 1,3-enyne compounds containing several substitution patterns was investigated. We initially studied the 1,3enyne with a terminal double bond and an internal triple bond such as 1-octen-3-yne (1a). The reaction was initiated by the addition of 2.0 equivalents of methanol to the mixture of 1a, 1.5 equivalents of bis(pinacolato)diboron 2, and 5 mol % of Cu(OtBu)/xantphos in THF at room temperature (Table 1,

Table 1: Monoborylation of 1,3-enyne compounds bearing a terminal double bond.[a]

Entry	Substrate	Ligand	Product	Yield [%] <sup>[b]</sup>	3/4 <sup>[c]</sup>
1 2 <sup>[d]</sup> 3 <sup>[d]</sup> 4 <sup>[d]</sup> 5	Bu 1a	xantphos dppe dppbz PPh <sub>3</sub> none	(pin)B Bu	87 61 60 80 0	> 95:5 > 95:5 > 95:5 > 95:5
6	cHex	xantphos	(pin)B 3b	88	> 95:5
7	1b	PPh <sub>3</sub>		89	> 95:5
8	Ph	xantphos	(pin)B 3c	83	94:6
9	1c	PPh <sub>3</sub>		79	> 95:5

[a] Reaction conditions: 1 (0.25 mmol), 2 (0.275-0.375 mmol), Cu-(OtBu) (5 mol%, 0.0125 mmol), ligand (5 mol%, 0.0125 mmol), THF (0.25 mL), and methanol (0.5 mmol). [b] Yield of isolated product. [c] Determined by <sup>1</sup>H NMR or GC analysis of the crude reaction mixture. [d] Yield based on <sup>1</sup>H NMR analysis of the crude reaction mixture. dppbz = 1,2-bis(diphenylphosphonio) benzene, dppe = 1,2-bis(diphenyl-phosphonio)phosphino)ethane, pin = pinacolato.

entry 1). The reaction was complete within 2 hours and gave 3-alkynylboronate **3a** in 87% yield with high regioselectivity (3/4 > 95.5). This reaction is the first example of the type I hydroboration of 1,3-enynes. Reactions using other diphosphine ligands such as dppe and dppbz resulted in lower yields (60-61%; Table 1, entries 2 and 3). The reaction with PPh<sub>3</sub> also afforded 3a in high yield (80%; Table 1, entry 4). In the absence of the ligand, the reaction did not proceed (Table 1, entry 5). [13] The reaction of 1,3-enynes with cHex or Ph groups at the 4-position proceeded to furnish the corresponding 3alkynylboronates **3** selectively (79-89%, 3/4 = 94:6 to > 95:5;Table 1, entries 6–9).

We next investigated the reaction of 1,3-enyne compounds with other substituent patterns. Interestingly, by changing the ligand the reaction with 1,3-envnes bearing 1-

Table 2: Monoborylation of 1,3-enyne compounds with 1- or 2-substitution.[a]

Entry	Substrate	Ligand	Product	Yield [%] <sup>[b]</sup>	3/5 <sup>[c]</sup>
<b>1</b> <sup>[d]</sup>	Me Bu	xantphos	Me Bu (pin)B 3d	65	>99:1
2 <sup>[e]</sup>	1d	PPh <sub>3</sub>	Bu <b>5d</b>	64	7:93
3	Bu	xantphos	Me Bu (pin)B 3d	58	>99:1
4	Me 1e	PPh <sub>3</sub>	Me Bu <b>5e</b>	66	1:>99
5 <sup>[f]</sup>	10	xantphos	3 d	61	> 99:1
6 <sup>[f]</sup>		PPh <sub>3</sub>	5 e	80	1:>99
7 <sup>[e]</sup>	Ph Me 1f	xanphos	(pin)B Ph Me 3f	52	92:8
<b>8</b> <sup>[e]</sup>		PPh <sub>3</sub>	Me Ph 5f	65	1:>99

[a] Reaction conditions: 1 (0.25 mmol), 2 (0.375 mmol), Cu(OtBu) (5 mol%, 0.0125 mmol), ligand (5 mol%, 0.0125 mmol), THF (0.25 mL), and methanol (0.5 mmol). [b] Yield of isolated product. [c] Determined by <sup>1</sup>H NMR or GC analysis of the crude reaction mixture. [d] Reaction time was 2.8 h. [e] 1.1 equivalents of diboron 2 was used. [f] 5 mol% of CuCl and 50 mol% of K(OtBu) were used instead of Cu(OtBu).

or 2-monosubstitution around the double bond afforded either 3-alkynylboronate 3 or 1,3-dienylboronate 5 selectively (Table 2).<sup>[14]</sup> With the xantphos ligand, the reaction of 1substituted 1,3-envne 1d afforded the corresponding 3alkynylboronate 3d with excellent regioselectivity (3/5> 99:1; Table 2, entry 1). In contrast, the reaction with PPh<sub>3</sub> gave 1,3-dienylboronate 5d with high regioselectivity (3/5= 7:93; Table 2, entry 2). The reaction with an (E)-alkene substrate also gave either 3-alkynylboronate 3d and 1,3dienylboronate 5e, respectively, thus demonstrating that the alkene geometry (E or Z) did not affect the selectivityoutcome (Table 2, entries 3 and 4). This reaction was also performed with the easily available CuCl/K(OtBu) precatalyst instead of Cu(OtBu) (Table 2, entries 5 and 6). This same type of product profile was also observed in the reaction with 2-substituted 1,3-enyne **1 f** (Table 2, entries 7 and 8).

We further tested the reaction of 1,3-enyne compounds with di- or trisubstitution around the double bond (Table 3). 1-Propynylcyclohexene 1g was converted into the corresponding 1,3-dienylboronate 5g in quantitative yield and with high regioselectivity (5/6 > 95:5; Table 3, entry 1). Other possible regioisomers, such as 3-alkynylboronate or 1,2dienylboronate were not detected. Using other bidentate ligands afforded 5g in high regioselectivity; however, the vields were lower when compared with the reaction using

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8

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Table 3: Monoborylation of 1,3-enyne compounds bearing 1,1-di-, 1,2-di-, and 1,1,2-trisubstitution.[a]

58

.OMe

89:11

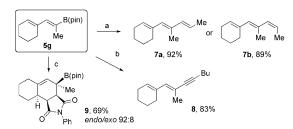
xantphos

1m

[a] Reaction conditions: 1 (0.25 mmol), 2 (0.275-0.5 mmol), Cu(OtBu) (5 mol%, 0.0125 mmol), ligand (5 mol%, 0.0125 mmol), THF (0.25 mL), and methanol (0.5 mmol). [b] Yield of isolated product. [c] Determined by <sup>1</sup>H NMR or GC analysis of the crude reaction mixture. [d] Reaction was carried out on a 0.5 mmol scale. [e] 5 mol% of CuCl and 50 mol% of K(OtBu) were used instead of 5 mol% of Cu(OtBu). [f] Catalyst loading was 10 mol %. [g] Yield based on <sup>1</sup>H NMR analysis of the crude reaction mixture.

xantphos (77–85%; Table 3, entries 2 and 3). The reaction with PPh<sub>3</sub> gave excellent yield and regioselectivity (97%, 5/6 > 95.5; Table 3, entry 4), but no reaction was observed in the absence of the ligand (Table 3, entry 5). The CuCl/ K(OtBu) precatalyst was also operative without significant loss of regioselectivity and yield (89%, 5/6 > 95:5; Table 3, entry 6). The reaction of a 1,3-enyne bearing 1,1-disubstitution (1h) afforded the corresponding 1,3-dienylboronate 5h with high regioselectivity (5/6 = 92:8; Table 3, entry 7). This reaction was applicable to the 1,3-enynes with an ether or a benzyloxy functionality (Table 3, entries 8 and 9). Reactions with 1,3-enynes bearing a terminal alkyne moiety also afforded the corresponding 1,3-dienylboronate 5k with high selectivities (5/6 > 95:5); Table 3, entries 10 and 11). The reaction of a 1,3-envne bearing 1,1,2-trisubstitution proceeded with high regioselectivity but the yield was lower even in the presence of 10 mol% of the catalyst (16%; Table 3, entry 12). This outcome was probably a result of the large steric hindrance around the double bond. Interestingly, the reaction of a 1,3-enyne with a phenyl group at the 4-position (1m) predominantly gave the type VI regioisomer (6m) with good selectivity (5/6 =15:85; Table 3, entry 13).

The usefulness of 1,3-dieneylboronate 5g was also demonstrated (Scheme 3). Palladium-catalyzed cross-cou-



Scheme 3. Derivatization of 1,3-dienylboronate (5 g). Reaction conditions: path a) (E)- or (Z)-1-bromopropene,  $[Pd(PPh_3)_4]/SPhos$ (cat.), aq NaOH (2 м), 60 °C, 2.5-3 h; path b) 1-iodohexyne, [Pd-(PPh<sub>3</sub>)<sub>4</sub>] (5 mol%), aq NaOH (2 м), 80 °C, 4 h; path c) N-phenylmareimide, 120 °C, 3 days. SPhos = 2-dicyclohexylphosphino-2',6'dimethoxybiphenyl.

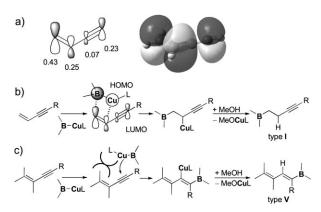
pling of  $\mathbf{5g}$  with (E)-, (Z)-1-bromopropene, and 1-iodohexyne gave the corresponding trienes and dienyne in high yields (7a: 92%, 7b: 89%, and 8: 83%; Scheme 3, path a,b). Furthermore, the Diels-Alder reaction<sup>[15]</sup> with N-phenylmaleimide afforded the unprecedented cyclic allylboronate 9, which has four contiguous stereocenters, including one quaternary carbon atom, with high diastereoselectivity (69%, endo/exo = 92:8; Scheme 2, path c).

The reaction for the asymmetric synthesis of enantioenriched 3-alkynylboronate with a chiral copper(I) catalyst (5 mol % of Cu(OtBu)/(R,R)-quinoxP\*) resulted in a good ee value with excellent regioselectivity (84 % ee, 3/5 = 95:5; Scheme 4); however, the yield was moderate (34%)

Scheme 4. Asymmetric catalytic monoborylation of 1,3-enyne 1 d.

because of the formation of multiborylation by-products. This reaction is the first example of an asymmetric synthesis of 3-alkynylboronates.

A tentative explanation of the regioselective outcome of the hydroboration reaction is presented in Scheme 5. According to the mechanistic investigation reported by Marder, Lin, and co-workers, [16] the interaction between the HOMO of the borylcopper intermediate and the electrophile LUMO is decisive in the regioselectivity of the borylcopper addition to unsaturated bonds. Orbital population analysis showed that



**Scheme 5.** a) DFT population analysis and isosurface of the LUMO of 1-buten-3-yne (B3LYP/6-31G(d,p), Gaussian 09W). b,c) Proposed explanation for product selectivities.

the 2p orbitals of the alkene carbon atoms (Scheme 5a; C1: 0.43; C2: 0.25 for 1-buten-3-yne) have a significantly larger contribution than those of the alkyne carbon atoms (C3: 0.07; C4: 0.23).

Thus, it is reasonable that the borylcupration takes place at the olefin double bond when steric perturbation of the substrate is not significant (type I; Scheme 5b). In the case of highly substituted 1,3-enynes, steric hindrance around the double bond would render borylcopper addition to the electronically less favorable alkyne moiety to produce 1,3-dienylboronates (type V; Scheme 5c). However, the selectivity outcome observed for moderately substituted 1,3-enynes were counterintuitive along this line. Further investigation is required to resolve this point.

In summary, we have developed copper(I)-catalyzed regioselective monoborylation of 1,3-enyne compounds. This catalysis includes the first examples for type I and VI hydroborations, and efficient type V hydroboration of 1,3-enynes with an internal triple bond. The regioisomeric preference (1,3-dienylboronates or 3-alkynylboronates) was primarily determined by the substrate structure, whereas the regioselectivity for moderately substituted 1,3-enynes was controlled by the ligand of the catalyst. It is noted that these selectivity features are different from those for 1,3-dienes in our previous study. [10] This copper(I)-catalyzed selective monoborylation is a complementary method to conventional hydroboration reactions for 1,3-enynes.

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 $\label{eq:Noncatalyzed reaction: Polyandary} \begin{array}{ll} \textbf{Noncatalyzed reaction:} & \textbf{B} = B(cat), \, neat, \, 70^{\circ}\text{C}, \, 15 \, h \\ \textbf{Rhodium-catalyzed reaction:} & \textbf{B} = B(pin), \, [RhCl(PPh_3)_3] \, (5 \, mol \, \%) \\ \textbf{CH}_2\text{Cl}_2, \, RT \, to \, 50^{\circ}\text{C}, \, 20 \, h \\ \end{array}$ 

75% yield, **5/6** = 76:24

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